Studies on fused pyrimidine derivatives. Part 14.¹ Formation and transformation of [4 + 2] cycloadducts, cyclohepta[g]quinazoline derivatives, by the reaction of 5-(arylaminomethylene)-1,3-dimethyl-6-methyleneperhydropyrimidine-2,4-diones with tropone

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The reaction of 5-(arylaminomethylene)-1,3-dimethyl-6-methyleneperhydropyrimidine-2,4-diones 3 with tropone 4 gave two diastereoisomeric [4 + 2] cycloadducts, cyclohepta[g]quinazolines 5 and 6. These products correspond to *endo*- and *exo*-approach of enamine 3 to the 2,3-double bond of tropone 4, respectively. The chemical behaviour of the *endo*- and *exo*-adducts will be discussed. The X-ray structures of 5-(4-bromoanilino)-1,3-dimethyl-2,3,4,*r*-5,*c*-5a,10,*c*-10a,11-octahydro-1*H*-cyclohepta[g]quinazoline-2,4, 10-trione, 5-(4-bromoanilino)-1,3-dimethyl-2,3,4,*r*-5,*t*-5a,10,*t*-10a,11-octahydro-1*H*-cyclohepta[g]quinazoline-2,4, 10-trione and 12,14-Dimethyl-*r*-6b,9,10,*c*-10a,11,12,13,14,15,*c*-15b,*c*-15c,16-dodecahydrobenzo[c]-cyclohepta[kl]pyrimido[5,4-h]acridine-10,13,15-trione are described.

In previous papers,² we described a generation of 5-(substituted aminomethylene)-1,3-dimethyl-6-methyleneperhydropyrimidine-2,4-dione intermediates **3** via a thermal 1,5-hydrogen shift of the corresponding 5-(substituted iminomethyl)-1,3,6-trimethyl-1,2,3,4-tetrahydropyrimidine-2,4-diones **2**. The intermediates **3** are regarded as a 1,3-diene located at the periphery of a heterocyclic system and underwent a single and regio- and stereo-selective [4 + 2] cycloaddition reaction with olefinic dienophiles to afford quinazoline derivatives (Scheme 1).



Scheme 1 *Reagents and conditions*: i, RNH₂, toluene or benzene, reflux; ii, 1,5-hydrogen shift; iii, methyl or ethyl acrylate (CH₂=CHZ)

Tropone 4 has emerged as one of the most typical multiple π electron systems (*i.e.*, 2-, 4-, 6-, and 8π -systems) in the higher order cycloaddition reactions. Thermal reaction of simple dienes (4π) with tropone **4** proceed usually in the [4 + 6] manner utilising the 6π -component of tropone.³ The [4 + 6] cycloaddition reaction, however, was suppressed by nonhydrogen substituents at the reaction sites, and therefore the [2 + 4] cycloaddition reaction predominated utilising the 2π component of the diene and the 4π -component of tropone instead.^{3c}

Only three examples were found for the [4 + 2] cycloaddition

reaction of diene (4π) and tropone (2π) ; *o*-xylylene cycloadded to the 4,5-double bond of tropone to give a cyclohepta[*b*]naphthalene derivative together with the [4 + 6] cycloadduct as a minor product.⁴ On the other hand, isobenzofuran reacted with tropone at its 2,3-double bond as well as giving the [4 + 6] cycloadduct.⁵

Tetrachlorocyclopentadienone ethylene ketal reacted with tropone to yield a 2:1 adduct resulting from double [4 + 2] cycloadditions on the 2,3- and 6,7-bond of tropone.⁶

We examined, therefore, the reaction of dienes 3 with tropone 4 in order to obtain further information on the reaction features of the diene.³ The reaction of 5-(arylamino methylene)-1,3-dimethyl-6-methyleneperhydropyrimidine-2,4-diones 3 with tropone 4 gave two diastereoisomeric [4 + 2] cycloadducts, cyclohepta[g]quinazolines 5 and 6, in moderate total yields. The cycloaddition of dienes 3 occurred regioselectively on the 2,3-double bond in tropone 4 and the adducts corresponded to endo- and exo-approaching products, respectively. These endo- and exo-[4 + 2] cycloadducts exhibited different behaviour toward acid treatment.

Results and discussion

[4 + 2] Cycloaddition of 5,6-dimethyleneperhydropyrimidine-2,4-diones 3 with tropone 4

The reaction of 5-anilinomethylene-1,3-dimethyl-6-methyleneperhydropyrimidine-2,4-dione **3a**, formed *in situ* by the condensation and successive isomerisation of 1,3,6-trimethyl-2,4-dioxo-1,2,3,4-tetrahydropyrimidine-5-carbaldehyde **1** and aniline, with tropone **4** in 1,4-dioxane under reflux for 6 h gave two 1:1 adducts **5a** and **6a** in 26 and 25% yield, respectively, together with the formation of polymeric products (Scheme 2). The adducts **5a** and **6a** were shown to be primary products by the reaction of diene **3a** and with tropone **4** as follows; a similar reaction in benzene or tetrahydrofuran (THF) under reflux for 6 h gave almost a 1:1 mixture of **5a** and **6a** (Table 1) and no interconversion between **5a** and **6a** was observed under the above conditions.

The structures 5a and 6a were elucidated on the basis of spectral data. The ¹³C NMR spectra of products 5a and 6a were in accord with each other over the sp³- and sp²-carbon ranges

 Table 1
 Reaction of 5-(arylaminomethylene)-1,3-dimethyl-6-methyleneperhydropyrimidine-2,4-diones 3 with tropone 4 (Scheme 2)

Entry		Ar	Solvent	Time (t/h)	Products (Yield ^a /%)		
	1	Ph	1,4-dioxane	6	5a (26)	6a (25)	
	2	Ph	benzene	6	5a (22)	6a (19)	
	3	Ph	THF	6	5a (14)	6a (13)	
	4	4-BrC ₆ H ₄	1,4-dioxane	6	5b (19)	6b (20)	
	5	4-MeOC ₆ H ₄	1,4-dioxane	6	5c (17)	6c (23)	
	6	3-MeOC H	1.4-dioxane	6	5d (24)	6d (22)	
	7	3-MeC ₆ H ₄	1,4-dioxane	6	5e (31)	6e (29)	
	8	1-Naphthyl	1.4-dioxane	6	5f (15)	6f (27)	7f (12)
	9	1-Naphthyl	1.4-dioxane	36	()	6f (25)	7f(27)
1	0	2-Naphthyl	toluene	6	5g and 7g (22) ^b	6g (25)	(27)
1	1	2-Naphthyl	1,4-dioxane	24		6g (15)	7g (15)

^a Isolated yield. ^b Combined yield because tricycle 5g was partly isomerised to hexacycle 7g during purification.



Scheme 2 Reaction of 5-(arylaminomethylene) diones 3 with tropone 4 in some refluxing solvents

except for only one sp³-carbon signal, assigned to C-5a. The ¹H-¹H COSY spectra of compounds 5a and 6a revealed that both adducts had an alignment of methine (5-H), methine (5a-H), methine (10a-H), and methylene protons (11-H) as well as of olefin (9-H), olefin (8-H), olefin (7-H), olefin (6-H), and methine proton (5a-H). These findings indicate that the cycloaddition of diene 3a occurs on the 2,3-double bond of tropone 4 and that the 5-(anilinomethylene)moiety of compound 3a is orientated in the opposite direction to that of the tropone carbonyl group. The stereochemistry of the adducts 5a and 6a was assigned from the coupling constants between 5and 5a-H; for adduct 5a the methine proton at the 5-position was observed as a broad singlet ($J \sim 0$ Hz, 5,5a-cis), while that for 6a appeared as a doublet (J 2.9 Hz, 5,5a-trans). The assignments were elucidated by nuclear overhauser enhancement (NOE) measurements of compounds 5a and 6a; irradiation of the 5a-H caused a remarkable enhancement of the 5-H signal for 5a (18%) and a little for that for compound 6a (1%). These suggest that tropone 4 adds formally to diene 3a with a Z-configuration 2^{e} at the 5-methylene moiety in endo- and exo-manner to form 5-anilino-1,3-dimethyl-2,3,4,5,5a,10,10a,11-octahydro-1*H*-cyclohepta[g]quinazoline-2,4,10-triones 5a and 6a, respectively. More details on the reaction pathway will be discussed in the following paper.⁷

Similar reactions of 5-(4-bromoanilinomethylene)- **3b**, 5-(4methoxyanilinomethylene)- **3c**, 5-(3-methoxyanilinomethylene)- **3d** and 5-(3-methylanilinomethylene)-substituted dione **3e** with tropone 4 gave mixtures of *endo*- and *exo*-adducts **5b**e and **6b**-e in moderate total yields (Table 1). The structures of these adducts **5** and **6** were also assigned on the basis of their spectral data. Furthermore the structures of adducts **5b** and **6b** were unambiguously established by X-ray crystallographic analyses (see the Experimental section).

A similar reaction of 5-(1-naphthylaminomethylene)-substituted dione 3f with tropone 4 in 1,4-dioxane under reflux for 6 h gave a mixture of three isomeric products 5f (15%), 6f (27%), and 7f (12%) (Scheme 3). The proportions of the products



Scheme 3 Isomerisation of *endo*-cycloadducts 5 to acridines 7 in acid conditions (H^+ or Lewis acids)

depended upon the reaction conditions; prolonged heating (36 h) in 1,4-dioxane gave compounds 6f (25%) and 7f (27%). These results suggested that product 7f was a secondary one from *endo*-adduct 5f and, indeed, heating of compound 5f in 1,4-dioxane under reflux gave compound 7f (Table 2).

The structure of compound **7f** was assigned to be a benzo[c]cyclohepta[k/]pyrimido[5,4-h]acridine derivative on the basis of its spectral data, and was also confirmed by X-ray crystallographic analysis (see the Experimental section). These suggest that nucleophilic attack of the naphthalene π -electrons on the δ -position of the seven-membered $\alpha,\beta,\gamma,\delta$ -unsaturated ketone moiety in compound **5f** takes place to give a hexacyclic system containing a seven-membered non-conjugated ketone (δ_{CO} 208.7).

Similar results were obtained by the reaction of 5-(2naphthylaminomethylene)-substituted dione 3g with tropone 4 in toluene or 1,4-dioxane under reflux which led to adducts 5g, 6g, and benzo[a]cyclohepta[kl]pyrimido[5,4-h]acridine 7g (Scheme 2). Such transformation of tricycles 5 to hexacycles 7 was consistent with the results of PM3 calculations as described latter.

Table 2 Thermal and acid-assisted isomerisation of endo-adducts 5 to acridines 7

 Entry	Substrate	Solvent	Acid catalyst	Temp.	Time (t/h)	Products (Yield ^a /%)
1 2 3 4 5 6 7	5f 5f 5g 5a 5d 5d 5d	1,4-dioxane EtOH EtOH EtOH EtOH CH ₂ Cl ₂ EtOH	none none 12 mol dm ⁻³ HCl (1 drop) 0.5 mol dm^{-3} HCl (1 drop) AlCl ₃ (0.2 mol equiv.) 0.5 mol dm^{-3} HCl (1 drop)	reflux reflux reflux reflux reflux ambient reflux	6 24 6 2 4 12 17	5f (42), 7f (42) 7f (91) 7g (quant.) 7a (29) 7d (61) 7d (86) 7e (52) ^b
8	5e	benzene	$BF_3(OEt_2)$ (0.1 mol equiv.)	reflux	1	7e (40) ^{<i>n</i>}

^a Isolated yield. ^b Mixture (~1:1) of two isomers by its ¹H NMR spectrum.

The diene 3 added to the 2,3-double bond of tropone 4 regioselectively to afford *endo*- and *exo*-[4 + 2] cycloadducts 5 and 6 as mentioned above. The tropone 4, therein, behaved only as a 2π -component toward the diene 4π -system of compounds 3. This is probably the first example reported of the thermal reactions of tropone with 4π -dienes.

Chemical behaviours of *endo*- and *exo*-[4 + 2] cycloadducts

In endo-adducts 5f and 5g, the π -electrons of the naphthylamino moiety at the 5-position underwent a nucleophilic ring closure to afford hexacyclic heterocycles 7f and 7g under the reaction conditions. Our next concern, therefore, was focused on the chemical behaviour of the [4 + 2] cycloadducts 5 and 6. Other endo-adducts 5 than compounds 5f and 5g did not show any change under the reaction conditions; neither did any of the exo-[4 + 2] cycloadducts 6. The isomerisation of tricycle 5f to hexacycle 7f was shown to proceed smoothly in ethanol under reflux (Table 2). Utilisation of proton or Lewis acids facilitated the isomerisation $5 \longrightarrow 7$ (Scheme 3); treatment of compound 5a with one drop of 12 mol dm⁻³ hydrochloric acid in ethanol under reflux for 2 h gave the acridine 7a in 29% yield. Similarly, acridines 7d and 7e (as two regioisomers) were obtained by the treatment of endo-adducts 5d and 5e with protic and Lewis acids. The results of the transformation $5 \longrightarrow 7$ upon acid treatment are summarised in Table 2.

On the other hand, treatment of *exo*-adducts **6a**, **6d** and **6e** with Lewis acids under similar conditions gave intractable mixtures of products. Prolonged heating of compounds **6a**, **6d** and **6e** in ethanol containing a higher concentration of hydrochloric acid led to 1,3-dimethyl-2,3,4,10-tetrahydro-1*H*-cyclohepta[g]quinazoline-2,4,10-trione **8** in low to fair yields along with mixtures of unidentified products (Scheme 4). The results of conversion $6 \longrightarrow 8$ are demonstrated in Table 3. 1,4-Elimination of the arylamine, accompanied by dehydrogenation, from substitute **5** was postulated for the formation of cycloheptaquinazoline **8**.



Scheme 4 Reagent and conditions: i, EtOH, hydrochloric acid, reflux

In order to obtain further information on the chemical behaviour of [4 + 2] cycloadducts 5 and 6, PM3 molecular orbital calculations were examined. The structure optimisations of adducts 5a and 6a were performed by utilising the structures

 Table 3
 Acid treatment of exo-adducts 6 leading to the cycloheptaquinazoline 8

Entry	Substrate	Acid catalyst	Time (t/h)	Product (Yield ^a /%)
1	6a	$12 \text{ mol } \text{dm}^{-3} \text{ HCl}$ (10 drops)	24	8 (29)
2	6d	12 mol dm ⁻³ HCl (1 drop)	20	8 (48)
3	6e	12 mol dm ⁻³ HCl (1 drop)	20	8 (40)



7a -54.31 kcal mol⁻¹



Fig. 1 Heats of formation for acridine derivatives 7a and 7a'. 1 cal = 4,184 J.

7a' -51.61 kcal mol⁻¹

of compounds **5b** and **6b** as initial geometries. These revealed that the isomerisation of *endo*-adduct **5a** to hexacycle **7a** was a possible process and that, on the other hand, the nucleophilic attack of the anilino π -electrons in compound **6a** on the sevenmembered unsaturated ketone moiety was impossible on structural grounds. The formation of the seven-membered β , γ unsaturated ketone system in compounds **7** was also confirmed by the PM3 calculations; the heat of formation for compound **7a** (β , γ -unsaturated) was lower than that estimated for the regioisomer **7a**' (α , β -unsaturated) by 2.7 kcal mol⁻¹⁺ (Fig. 1).

In conclusion, we have reported here that 5-(arylamino methylene)-6-methyleneperhydropyrimidine-2,4-dione intermediates 3 cycloadd to the 2,3-double bond of tropone 4 regioselectively to afford *endo*- and *exo*-[4 + 2] cycloadducts 5 and 6. In *endo*-adduct 5 the π -electrons of the arylamino moiety attack the δ -position of $\alpha,\beta,\gamma,\delta$ -unsaturated ketone system derived from the tropone to afford the acridine 7 under both neutral and acidic conditions.

Experimental

Mps were measured on a Yanagimoto micro melting point apparatus and are uncorrected. IR spectra were measured on a JASCO IR-Report-100 spectrophotometer from samples as

 $\dagger 1 \text{ cal} = 4.184 \text{ J}.$

KBr pellets. ¹H NMR and ¹³C NMR spectra were measured on JEOL GSX-400 and/or 270 spectrometers for solutions in deuteriochloroform unless otherwise stated. Tetramethylsilane was used as internal standard and *J*-values are given in Hz; Splitting pattern ov indicates signals overlapping with each other. Mass spectra were determined on a JEOL JMS-021G-2 or JMS-D spectrometer. Elemental analyses were performed on a Hitachi 026 CHN analyser. All non-aqueous reactions were run under a positive pressure of argon. All solvents were dried by standard methods before use. The progress of reactions was monitored by TLC (Silica Gel 60F-254, Merck). Chromatographic purification was performed with Wakogel C-200 (100–200 mesh, Wako Pure Chemical Industries) and/or Silica Gel 60 (230–400 mesh, Merck).

Reaction of 5-anilinomethylene-1,3-dimethyl-6methyleneperhydropyrimidine-2,4-dione 3a with tropone 4. General procedures

To a refluxing solution of tropone 4 (1.16 g, 10.0 mmol) in 1,4dioxane (5 cm³) were added solutions of 1,3,6-trimethyl-2,4dioxo-1,2,3,4-tetrahydropyrimidine-5-carbaldehyde 1 (0.182 g, 1.0 mmol) and aniline (0.109 cm³, 1.3 mmol) in 1,4-dioxane (2.5 cm³ each) via a double-barrelled microfeeder over a period of 3 h. The reaction mixture was heated under reflux for an additional 6 h and the solvent was then evaporated off under reduced pressure. The residue was subjected to column chromatography on silica gel with hexane-ethyl acetate (1:1) to give a mixture of the [4 + 2] cycloadducts 5a and 6a (0.188 g; 5a:6a ~ 1:1). Flash chromatography of the mixture with hexane-ethyl acetate (2:1) gave endo- 5a (26%) and exo-adduct 6a (25%), respectively.

5-Anilino-1,3-dimethyl-2,3,4,r-5,c-5a,10,c-10a,11-octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **5a** was obtained as orange plates from ethanol; mp 205–207 °C (Found: C, 69.2; H, 5.9; N, 11.4. C₂₁H₂₁N₃O₃ requires C, 69.40; H, 5.83; N, 11.56%); v_{max}/cm^{-1} 3350 (NH) and 1680 and 1640 (CO); δ_{H} (270 MHz) 2.49 (1 H, dd, J 5.9 and 17.6, 11-H), 2.86 (1 H, br s, 5a-H), 3.10 (1 H, br, NH), 3.31 and 3.52 (each 3 H, each s, 1- and 3-Me), 3.43 (1 H, d, J 17.6, 11-H), 3.67 (1 H, m, 10a-H), 4.81 (1 H, br, 5-H), 5.89 (1 H, dd, J 7.8 and 11.7, 7-H), 6.00 (1 H, d, J 12.2, 9-H), 6.30–6.39 (2 H, ov, 6- and 8-H) and 6.56, 6.65 and 7.08 (total 5 H, Ph); δ_c (68 MHz) 27.4 (C-11), 28.2 (1-Me), 31.2 (3-Me), 43.4 (C-5a), 44.4 (C-10a), 49.4 (C-5), 108.4 (C-4a), 113.5, 117.7, 128.8 and 148.1 (Ph-C), 126.9 and 129.9 (C-7 and -9), 136.4 and 138.4 (C-6 and -8), 149.0 (C-11a), 151.9 (C-2), 161.5 (C-4) and 201.6 (C-10); m/z 364 (M⁺).

5-Anilino-1,3-dimethyl-2,3,4,r-5,t-5a,10,t-10a,11-octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **6a** was obtained as yellow plates from ethanol; mp 208–210 °C (Found C, 69.7; H, 5.7; N, 11.6%); v_{max} /cm⁻¹ 3380 (NH) and 1700, 1650 and 1640 (CO); $\delta_{\rm H}$ (270 MHz) 2.62–2.72 (2 H, ov, 11-H₂), 3.21–3.28 (3 H, ov, 5a- and 10-H and NH), 3.28 and 3.39 (each 3 H, each s, 1and 3-Me), 4.59 (1 H, d, J 2.9, 5-H), 5.89 (1 H, d, J 11.2, 6-H), 6.08 (1 H, d, J 12.7, 9-H), 6.17 (1 H, ddd, J 2.9, 7.3 and 11.2, 7-H), 6.60–6.65 (4 H, ov, 8-H and Ph) and 7.12 (2 H, Ph); $\delta_{\rm C}$ (68 MHz) 23.3 (C-11a), 26.7 (1-Me), 29.5 (3-Me), 33.5 (C-5a), 44.1 (C-10a), 49.4 (C-5), 105.9 (C-4a), 111.2, 115.6, 125.9 and 145.8 (Ph-C), 127.9 and 129.2 (C-7 and -9), 136.4 and 138.1 (C-6 and -8), 149.7 (C-11a), 150.1 (C-2), 160.4 (C-4) and 199.0 (C-10); *m*/z 364 (M⁺).

5-(4-Bromoanilino)-1,3-dimethyl-2,3,4,r-5,c-5a,10,c-10a,11octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **5b** was obtained as orange plates from ethanol; mp 206–208 °C (Found: C, 57.2; H, 4.6; N, 9.4. $C_{21}H_{20}BrN_3O_3$ requires C, 57.02; H, 4.57; N, 9.50%); v_{max}/cm^{-1} 3360 (NH) and 1690 and 1650 (CO); $\delta_H(270 \text{ MHz})$ 2.50 (1 H, dd, J 5.9 and 17.6, 11-H), 2.84 (1 H, m, 5a-H), 3.07 (1 H, br, NH), 3.30 and 3.52 (each 3 H, each s, 1- and 3-Me), 3.43 (1 H, d, J 17.6, 11-H), 3.69 (1 H, m, 10a-H), 4.75 (1 H, dd, J 3.3 and 7.3, 5-H), 5.89 (1 H, dd, J 7.7 and 11.7, 7-H), 5.99 (1 H, d, J 12.8, 9-H), 6.30–6.37 (2 H, ov, 6-and 8-H) and 6.44 and 7.15 (total 4 H, ArH); δ_{c} (68 MHz) 27.3 (C-11), 28.2 (1-Me), 31.1 (3-Me), 43.3 (C-5a), 44.1 (C-10a), 49.3 (C-5), 107.9 (C-4a), 109.2, 114.9, 131.5 and 147.2 (Ph-C), 136.2 and 138.0 (C-6 and -8), 149.1 (C-11a), 151.8 (C-2), 161.5 (C-4) and 201.6 (C-10); *m/z* 443 and 441 (M⁺).

5-(4-*Bromoanilino*)-1,3-*dimethyl*-2,3,4,r-5,t-5a,10,t-10a,11octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **6b** was obtained as pale yellow plates from ethanol; mp 200–203 °C (Found: C, 56.8; H, 4.6; N, 9.2%); ν_{max}/cm^{-1} 3350 (NH) and 1690 and 1650 (CO); $\delta_{\rm H}$ (270 MHz) 2.54–2.69 (2 H, ov, 11-H₂), 3.21–3.29 (2 H, ov, 5a-H and NH), 3.35 and 3.38 (each 3 H, each s, 1- and 3-Me), 3.83 (1 H, br d, J 4.0, 10a-H), 5.59 (1 H, br, 5-H), 5.89 (1 H, br d, J 13.2, 6-H), 6.13–6.21 (2 H, ov, 7- and 9-H), 6.61 (1 H, dd, J 7.0 and 12.5, 8-H) 6.52 and 7.25 (total 4 H, ArH); $\delta_{\rm C}$ (68 MHz) 24.8 (C-11), 28.3 (1-Me), 31.0 (3-Me), 35.0 (C-5a), 46.1 (C-10a), 51.7 (C-5), 107.2 (C-4a), 110.1, 114.9, 132.9 and 145.5 (Ph-C), 137.7 and 138.7 (C-6 and -8), 151.2 and 151.5 (C-2 and -11a), 162.0 (C-4) and 200.3 (C-10); *m/z* 443 and 441 (M⁺).

Structures of compounds **5b** and **6b** were confirmed by X-ray crystal-structure analyses and their crystal data are summarised in Table 4 (see below).

5-(4-Methoxyanilino)-1,3-dimethyl-2,3,4,r-5,c-5a,10,c-10a,11octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione 5c was obtained as orange plates from ethanol; mp 192-194 °C (Found: C, 67.4; H, 6.0; N, 10.8. C₂₂H₂₃N₃O₄ requires C, 67.16; H, 5.89; N, 10.68%); v_{max}/cm⁻¹ 3350 (NH) and 1690 and 1640 (CO); δ_H(270 MHz) 2.49 (1 H, dd, J 6.2 and 17.6, 11-H), 2.85 (1 H, br, 5a-H), 3.30 and 3.51 (each 3 H, each s, 1- and 3-Me), 3.40 (1 H, d, J 17.6, 11-H), 3.62 (1 H, br, 10a-H), 3.71 (3 H, s, OMe), 3.30-3.71 (1 H, br, NH), 4.69 (1 H, br, 5-H), 5.91 (1 H, dd, J 7.7 and 11.5, 7-H), 5.99 (1 H, d, J 12.5, 9-H), 6.30-6.40 (2 H, ov, 6- and 8-H) and 6.56 and 6.69 (total 4 H, ArH); $\delta_{\rm C}$ (68 MHz) 27.3 (C-11), 28.2 (1-Me), 31.1 (3-Me), 43.1 and 44.5 (C-5a and -10a), 50.9 (C-5), 55.7 (OMe), 108.7 (C-4a), 114.4, 115.2, 142.4 and 152.3 (Ph-C), 126.7 and 130.3 (C-7 and -9), 136.3 and 138.5 (C-6 and -8), 149.0 (C-11a), 151.9 (C-2), 161.6 (C-4) and 201.4 (C-10); m/z 393 (M⁺).

5-(4-*Methoxyanilino*)-1,3-*dimethyl*-2,3,4,r-5,t-5a,10,t-10a,11octahydro-1H-*cyclohepta*[g]*quinazoline*-2,4,10-*trione* **6c** was obtained as pale orange plates from ethanol; mp 192–194 °C (Found: C, 67.3; H, 6.0; N, 10.7%); v_{max} /cm⁻¹ 3380 (NH) and 1700, 1660 and 1640 (CO); δ_{H} (270 MHz) 2.53–2.71 (2 H, ov, 11-H₂), 3.30–3.38 (3 H, ov, 5a- and 10a-H and NH), 3.36 and 3.38 (each 3 H, each s, 1- and 3-Me), 3.73 (3 H, s, OMe), 4.56 (1 H, d, *J* 2.6, 5-H), 5.86 (1 H, d, *J* 11.4, 6-H), 6.11–6.18 (2 H, ov, 7- and 9-H), 6.58 (1 H, dd, *J* 7.0 and 12.5, 8-H) and 6.65 and 6.78 (total 4 H, ArH); δ_{C} (68 MHz) 24.9 (C-11), 28.2 (1-Me), 31.0 (3-Me), 35.1 (C-5a), 45.9 (C-10a), 52.9 (C-5), 55.8 (OMe), 107.8 (C-4a), 115.0, 115.6, 140.7 and 153.0 (Ph-C), 127.5 and 131.0 (C-7 and -9), 137.6 and 139.3 (C-6 and -8), 151.1 and 151.6 (C-2 and -11a), 162.2 (C-4) and 200.5 (C-10); *m/z* 393 (M⁺).

5-(3-*Methoxylanilino*)-1,3-*dimethyl*-2,3,4,r-5,c-5a,10,c-10a, 11-*octahydro*-1H-*cyclohepta*[g]*quinazoline*-2,4,10-*trione* 5d was obtained as yellow prisms from ethanol; mp 191–193 °C (Found: C, 67.5; H, 5.9; N, 10.5%); v_{max}/cm^{-1} 3350 (NH) and 1690, 1650 and 1640 (CO); δ_{H} (270 MHz) 2.49 (1 H, dd, *J* 6.2 and 17.6, 11-H), 2.86 (1 H, br, 5a-H), 3.13 (1 H, br, NH), 3.30 and 3.51 (each 3 H, each s, 1- and 3-Me), 3.41 (1 H, d, *J* 17.6, 11-H), 3.65 (1 H, br, 10a-H), 3.72 (3 H, s, OMe), 4.78 (1 H, br, 5-H), 5.91 (1 H, dd, *J* 7.7 and 11.4, 7-H), 5.99 (1 H, d, *J* 13.1, 9-H), 6.31–6.40 (2 H, ov, 6- and 8-H) and 6.11–6.24 and 6.99 (total 4 H, ArH); δ_{C} (68 MHz) 27.3 (C-11), 28.2 (1-Me), 31.1 (3-Me), 43.2 (C-5a), 43.2 and 44.5 (C-5a and -10a), 49.5 (C-5), 55.0 (OMe), 99.8, 102.8, 106.7, 129.4, 149.1 and 160.3 (Ph-C), 108.2 (C-4a), 126.8 and 129.9 (C-7 and -9), 136.5 and 138.3 (C-6) and -8), 149.5 (C-11a), 151.9 (C-2), 161.6 (C-4) and 201.4 (C-10); m/z 393 (M⁺).

5-(3-Methoxylanilino)-1,3-dimethyl-2,3,4,r-5,t-5a,10,t-10a,

11-octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione 6d was obtained as pale yellow plates from ethanol; mp 165– 167 °C (Found: C, 67.14; H, 5.80; N, 10.6%); v_{max}/cm^{-1} 3360 (NH) and 1700, 1690 and 1650 (CO); $\delta_{H}(270 \text{ MHz}) 2.53-2.72$ (2 H, ov, 11-H₂), 3.24–3.39 (2 H, ov, 5a- and 10a-H), 3.36 and 3.39 (each 3 H, each s, 1- and 3-Me), 3.75 (3 H, s, OMe), 4.64 (1 H, d, J 2.6, 5-H), 5.87 (1 H, d, J 3.3 and 12.5, 6-H), 6.12–6.32 (5 H, ov, 7- and 9-H and ArH), 6.59 (1 H, dd, J 7.0 and 12.5, 8-H) and 7.09 (1 H, t, J 8.1, ArH); $\delta_{C}(68 \text{ MHz})$ 24.9 (C-11), 28.3 (1-Me), 31.0 (3-Me), 35.1 (C-5a), 46.2 (C-10a), 51.6 (C-5), 55.1 (OMe), 99.7, 103.2, 106.4, 130.2, 147.8 and 160.8 (Ph-C), 107.5 (C-4a), 127.6 and 131.0 (C-7 and -9), 137.7 and 139.0 (C-6 and -8), 151.1 and 151.6 (C-2 and -11a), 162.0 (C-4) and 200.4 (C-10); m/z 393 (M⁺).

1,3-Dimethyl-5-(m-toluidino)-2,3,4,r-5,c-5a,10,c-10a,11-octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **5e** was obtained as orange prisms from ethanol; mp 199 °C (Found: C, 70.1; H, 6.2; N, 11.0. $C_{22}H_{23}N_3O_3$ requires C, 70.01; H, 6.14; N, 11.12%); v_{max}/cm^{-1} 3350 (NH) and 1690 and 1640 (CO); $\delta_{H}(270 \text{ MHz})$ 2.21 (3 H, s, Me), 2.49 (1 H, dd, J 5.9 and 18.0, 11-H), 2.87 (1 H, br, 5a-H), 3.08 (1 H, br, NH), 3.30 and 3.51 (each 3 H, each s, 1- and 3-Me), 3.40 (1 H, d, J 17.6, 11-H), 3.64 (1 H, br, 10a-H), 4.79 (1 H, br, 5-H), 5.90 (1 H, dd, J 8.1 and 11.7, 7-H), 5.99 (1 H, d, J 12.5, 9-H), 6.30–6.49 (5 H, ov, 6- and 8-H and ArH), 6.98 (1 H, dd, J 7.3 and 7.8, ArH); $\delta_{C}(68 \text{ MHz})$ 21.7 (Me), 27.3 (C-11), 28.2 (1-Me), 31.1 (3-Me), 43.2 (C-5a), 44.5 (C-10a), 49.6 (C-5), 108.4 (C-4a), 110.6, 114.5, 118.8, 128.5, 138.3 and 148.1 (Ph-C), 136.3 and 138.4 (C-6 and -8), 149.0 (C-11a), 151.9 (C-2), 161.5 (C-4) and 201.5 (C-10); m/z 377 (M⁺).

1,3-Dimethyl-5-(m-toluidino)-2,3,4,r-5,t-5a,10,t-10a,11-octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **6e** was obtained as yellow needles from ethanol; mp 214–215 °C (Found: C, 69.7; H, 6.2; N, 10.9%); v_{max}/cm^{-1} 3350 (NH) and 1690, 1650 and 1640 (CO); $\delta_{H}(270 \text{ MHz})$ 2.26 (3 H, s, Me), 2.49–2.70 (2 H, ov, 11-H₂), 3.25–3.36 (3 H, ov, 5a- and 10a-H and NH), 3.36 and 3.39 (each 3 H, each s, 1- and 3-Me), 4.66 (1 H, d, J 2.6, 5-H), 5.89 (1 H, br d, J 10.4, 6-H), 6.13–6.20 (2 H, ov, 7- and 9-H), 6.44–6.46 (2 H, ArH), 6.56–6.63 (2 H, ov, 8-H and ArH) and 7.07 (1 H, dd, J 7.3 and 7.7, ArH); $\delta_{C}(68 \text{ MHz})$ 21.7 (Me), 24.9 (C-11), 28.3 (1-Me), 31.0 (3-Me), 35.1 (C-5a), 46.1 (C-10a), 51.6 (C-5), 107.6 (C-4a), 110.4, 114.2, 119.4, 129.3, 139.1, and 146.5 (Ph-C), 127.6 and 131.0 (C-7 and -9), 137.6 and 139.1 (C-6 and -8), 151.0 and 151.6 (C-2 and -11a), 162.0 (C-4) and 200.4 (C-10); m/z 377 (M⁺).

1,3-Dimethyl-5-(1-naphthylamino)-2,3,4,r-5,c-5a,10,c-10a, 11-octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione 5f was obtained as orange crystals and was subjected to analytical treatment without recrystallisation; mp 160-162 °C (Found: C, 72.2; H, 5.7; N, 9.9. $C_{25}H_{23}N_3O_3$ requires C, 72.62; H, 5.61; N, 10.16%); v_{max}/cm^{-1} 3400 (NH) and 1700, 1650 and 1640 (CO); $\delta_{\rm H}(270 \text{ MHz})$ 2.52 (1 H, dd, J 5.9 and 17.6, 11-H), 2.96 (1 H, m, 5a-H), 3.27 and 3.53 (each 3 H, each s, 1- and 3-Me), 3.27-3.53 (2 H, ov, 11-H and NH), 3.71 (1 H, m, 10a-H), 5.09 (1 H, br, 5-H), 5.76 (1 H, dd, J7.7 and 11.7, 7-H), 5.89 (1 H, d, J 12.1, 9-H), 6.11 (1 H, dd, J 7.7 and 12.1, 8-H), 6.33 (1 H, dd, J 6.6 and 11.7, 6-H) and 6.85, 7.14-7.42 and 7.67-7.71 (total 7 H, ArH); $\delta_{c}(68 \text{ MHz})$ 27.4 (C-11), 28.1 (1-Me), 31.0 (3-Me), 43.4 and 44.5 (C-5a and -10a), 48.2 (C-5), 108.2 (C-4a), 106.0, 117.4, 120.5, 123.9, 124.5, 125.4, 126.1, 128.3, 134.4 and 143.1 (naphthyl-C), 126.5 and 130.0 (C-7 and -9), 136.1 and 138.4 (C-6 and -8), 149.6 and 151.8 (C-2 and -11a), 161.6 (C-4) and 201.8 (C-10); m/z 413 (M⁺).

1,3-Dimethyl-5-(1-naphthylamino)-2,3,4,r-5,t-5a,10,t-10a,11octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **6f** was obtained as yellow needles from ethanol-chloroform; mp 224225 °C (Found: C, 72.5; H, 5.6; N, 10.2%); ν_{max}/cm^{-1} 3400 (NH) and 1700, 1660 and 1640 (CO); $\delta_{H}(270 \text{ MHz})$ 2.58–2.78 (2 H, ov, 11-H₂), 3.41 and 3.51 (each 3 H, each s, 1- and 3-Me), 3.36–3.50 (3 H, ov, 5a- and 10a-H and NH), 4.88 (1 H, d, J 2.2, 5-H), 5.95 (1 H, br d, J 11.4, 6-H), 6.11–6.22 (2 H, ov, 7- and 9-H), 6.59 (1 H, dd, J 7.0 and 12.5, 8-H) and 6.76, 7.26–7.46 and 7.76–7.79 (total 7 H, naphthyl-H); $\delta_{C}(68 \text{ MHz})$ 25.1 (C-11), 28.3 (1-Me), 31.0 (3-Me), 34.3 (C-5a), 46.6 (C-10a), 51.7 (C-5), 107.4 (C-4a), 105.6, 118.8, 120.2, 124.1, 124.9, 125.8, 126.3, 128.6, 134.4 and 141.5 (naphthyl-C), 127.7 and 131.1 (C-7 and -9), 137.6 and 139.0 (C-6 and -8), 151.4 and 151.6 (C-2 and -11a), 162.2 (C-4) and 200.5 (C-10); m/z 413 (M⁺).

12,14-Dimethyl-r-6b,9,10,c-10a,11,12,13,14,15,c-15b,c-15c, 16-dodecahydrobenzo[c]cyclohepta[kl]pyrimido[5,4-h]acridine-10,13,15-trione 7f was obtained as pale brown plates from ethanol-dichloromethane; mp 241-243 °C (Found: C, 72.7; H, 5.6; N, 10.2%); v_{max}/cm⁻¹ 3380 (NH) and 1700, 1680 and 1660 (CO); $\delta_{\rm H}(270~{\rm MHz})$ 2.39 (1 H, dd, J 6.2 and 18.3, 11-H), 2.57 (1 H, dd, J 8.1 and 19.8, 9-H), 2.74-2.86 (2 H, ov, 9and 15c-H), 3.31 (1 H, dd, J 1.1 and 18.3, 11-H), 3.41 and 3.51 (each 3 H, each s, 12- and 14-Me), 3.78 (1 H, dd, J 6.2 and 6.4, 10a-H), 4.39 (1 H, d, J 3.7, 15b-H), 4.44 (1 H, t, J 9.9, 6b-H), 4.86 (1 H, br, NH), 5.91 (1 H, ddd, J 4.4, 8.1, and 10.4, 8-H), 6.59 (1 H, ddd, J 2.9, 9.9, and 10.4, 7-H), 7.15 (1 H, d, J 8.4, 5-H), 7.26 (1 H, d, J 8.4, 1-H), 7.37-7.44 (2 H, ov, 2- and 3-H) and 7.63–7.75 (2 H, ov, 4- and 6-H); $\delta_{\rm C}$ (68 MHz) 26.6 (C-11), 28.2 (12-Me), 31.0 (14-Me), 37.1 and 38.0 (C-9 and -15c), 41.5 and 43.8 (C-6b and -10a), 48.9 (C-15b), 107.3 (C-15a), 117.6 (C-6a), 119.4 and 119.9 (C-1 and -5), 122.8 (C-16b), 125.5 and 125.6 (C-2 and -3), 127.4, 128.2, 128.4 and 132.6 (C-4, -6, -7 and -8), 135.4 (C-4a), 140.8 (C-16a), 149.9 and 151.9 (C-11a and -13), 162.7 (C-15), and 208.7 (C-10); m/z 413 (M⁺).

The reaction mixture of 5-(2-naphthylamino)-substituted dione **3g** with tropone 4 was heated under reflux for 6 h. Usual work-up with short-column chromatography on silica gel gave a mixture of *endo*- **5g** and *exo*-adduct **6g** (in total yield of 47%). *endo*-Adduct **5g** was partly isomerised to the acridine **7f** during further purification.

1,3-Dimethyl-5-(2-naphthylamino)-2,3,4,r-5,c-5a,10,c-10a,11octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione 5g was obtained as orange crystals and was subjected to analytical treatment without recrystallisation; mp 203-205 °C (Found: C, 72.2; H, 5.8; N, 9.7%); v_{max}/cm⁻¹ 3380 (NH) and 1700, 1660 and 1650 (CO); $\delta_{\rm H}(\rm 270~MHz)$ 2.48 (1 H, dd, J 5.9 and 17.6, 11-H), 2.91 (1 H, m, 5a-H), 3.30 and 3.52 (each 3 H, each s, 1- and 3-Me), 3.42 (1 H, br d, J 17.6, 11-H), 3.30-3.52 (1 H, ov, NH), 3.67 (1 H, m, 10a-H), 4.96 (1 H, br, 5-H), 5.81 (1 H, dd, J 7.8 and 11.7, 7-H), 6.00 (1 H, d, J 12.2, 9-H), 6.26 (1 H, dd, J 7.8 and 12.2, 8-H), 6.40 (1 H, dd, J 6.8 and 11.7, 6-H) and 6.73, 6.90, 7.15, 7.31 and 7.52–7.62 (total 7 H, naphthyl-H); $\delta_{\rm C}$ (68 MHz) 27.4 (C-11), 28.3 (1-Me), 31.0 (3-Me), 43.4 and 44.5 (C-5a and -10a), 48.2 (C-5), 108.2 (C-4a), 106.2, 117.4, 120.5, 123.9, 124.5, 125.4, 126.1, 134.4 and 143.1 (naphthyl-C), 126.5 and 130.0 (C-7 and -9), 128.3 (C-3), 136.1 and 138.4 (C-6 and -8), 149.6 (C-11a), 151.8 (C-2), 161.6 (C-4) and 201.8 (C-10); m/z 413 (M⁺).

1,3-Dimethyl-5-(2-naphthylamino)-2,3,4,r-5,t-5a,10,t-10a,11octahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **6g** was obtained as yellow needles from ethanol–chloroform; mp 172– 174 °C (Found: C, 72.5; H, 5.8; N, 9.9%); v_{max} /cm⁻¹ 3350 (NH) and 1720, 1700 and 1650 (CO); δ_{H} (270 MHz) 2.53–2.70 (2 H, ov, 11-H₂), 3.24–3.34 (1 H, ov, 5a-H), 3.34 and 3.35 (each 3 H, each s, 1- and 3-Me), 3.46 (1 H, m, 10a-H), 3.98 (1 H, br, NH), 4.78 (1 H, d, J 2.2, 5-H), 5.93 (1 H, br d, J 11.4, **6**-H), 6.09–6.20 (2 H, ov, 7- and 9-H), 6.56 (1 H, dd, J 7.0 and 12.5, 8-H) and 6.83– 6.87, 7.20, 7.35 and 7.59–7.66 (total 7 H, naphthyl-H); δ_{C} (68 MHz) 24.8 (C-11), 28.2 (1-Me), 30.9 (3-Me), 34.7 (C-5a), 46.2 (C-10a), 51.6 (C-5), 107.2 (C-4a), 105.1, 118.2, 122.3, 126.0, 10,12-Dimethyl-r-4c,7,8,c-8a,9,10,11,12,13,c-13b,c-13c,14dodecahydrobenzo[a]cyclohepta[k1]pyrimido[5,4-h]acridine-8, 11,13-trione 7g was obtained as prisms from ethanoldichloromethane; mp 260 °C (decomp.) (Found: C, 72.4; H, 5.5; N, 9.85%); ν_{max}/cm^{-1} 3350 (NH) and 1690 and 1650 (CO); $\delta_{H}(CF_{3}CO_{2}D; 270 \text{ MHz}) 2.62$ (1 H, br d, J 19.5, 9-H), 3.09–3.24 (2 H, ov, 7-H₂), 3.47 (1 H, d, J 19.5, 9-H), 3.60 and 3.72 (each 3 H, each s, 10- and 12-Me), 3.60–3.84 (2 H, ov, 13c-H and NH), 4.54 (1 H, br, 8a-H), 5.08 (1 H, t, J 8.3, 4b-H), 5.28 (1 H, br s, 13b-H), 6.05 (1 H, m, 6-H), 6.93 (1 H, m, 5-H), 7.45 (1 H, d, J 8.8, 15-H), 7.71 (2 H, ov, 2- and 3-H), 8.01 (2 H, ov, 1- and 4-H) and 8.11 (1 H, br d, J 7.3, 16-H); m/z 413 (M⁺).

Conversion of endo-adducts 5 into acridines 7

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General procedures for treatment with hydrochloric acid. A solution of *endo*-adduct **5a** (0.100 g, 0.28 mmol) in ethanol (5 cm³) containing one drop of 12 mol dm⁻³ hydrochloric acid was heated under reflux for 12 h. Usual work-up with column chromatography gave the acridine **7a** (0.029 g, 29%).

10,12-Dimethyl-r-4b,7,8,c-8a,9,10,11,12,13,c-13b,c-13c,14dodecahydrocyclohepta[mn]pyrimido[4,5-c]acridine-8,11,13trione **7a** was obtained as plates from ethanol–chloroform; mp 268 °C (decomp.) (Found: C, 69.1; H, 5.85; N, 11.5. $C_{21}H_{21}N_3O_3$ requires C, 69.40; H, 5.83; N, 11.56%); v_{max} /cm⁻¹ 3350 (NH) and 1690, 1650 and 1645 (CO); δ_H (270 MHz) 2.39 (1 H, dd, J 7.3 and 18.3, 9-H), 2.66 (1 H, dd, J 7.3 and 19.8, 7-H), 2.77 (1 H, ddd, J 3.7, 5.9 and 9.5, 13c-H), 2.90 (1 H, ddd, J 2.9, 4.4 and 19.8, 7-H), 3.31 (1 H, d, J 18.3, 9-H), 3.37 and 3.50 (each 3 H, each s, 10- and 12-Me), 3.76 (1 H, dd, J 5.9 and 7.3, 8a-H), 3.88 (1 H, br, NH), 4.27–4.34 (2 H, ov, 4b- and 13b-H), 5.93 (1 H, ddd, J 4.4, 7.3 and 10.3, 6-H), 6.53–6.61 (2 H, ov, 1- and 5-H), 6.75 (1 H, t, J 7.3, 3-H) and 6.97–7.05 (2 H, ov, 2- and 4-H); m/z 363 (M⁺).

General procedures for treatment with lewis acids. A solution of *endo*-adduct 5d (0.050 g, 0.13 mmol) and aluminium chloride (0.004 g, 0.12 mmol) in dry dichloromethane (3 cm^3) was stirred at room temperature for 12 h. The reaction mixture was washed with water and the aqueous layer was extracted with dichloromethane. The combined organic layer was dried (MgSO₄) and evaporated to dryness. The residue was subjected to column chromatography on silica gel with hexane–ethyl acetate (1:2) which gave the acridine 7d (0.043 g, 86%).

2-Methoxy-10,12-dimethyl-r-4b,7,8,c-8a,9,10,11,12,13,c-13b, c-13c,14-dodecahydrocyclohepta[mn]pyrimido[4,5-c]acridine-8,11,13-trione 7d was obtained as pale yellow prisms from ethanol-chloroform; mp 231-233 °C; v_{max}/cm^{-1} 3340 (NH) and 1690, 1650 and 1640 (CO); $\delta_{\rm H}(270 \text{ MHz}) 2.39 (1 \text{ H},$ dd, J7.3 and 18.1, 9-H), 2.66 (1 H, dd, J7.8 and 19.5, 7-H), 2.74 (1 H, ddd, J 3.4, 6.4 and 9.8, 13c-H), 2.90 (1 H, ddd, J 2.9, 4.4 and 19.5, 7-H), 3.30 (1 H, dd, J 1.5 and 18.1, 9-H), 3.37 and 3.49 (each 3 H, each s, 10- and 12-Me), 3.71 (3 H, s, OMe), 3.74 (1 H, dd, J 6.4 and 7.3, 8a-H), 3.71-3.74 (1 H, br, NH), 4.23 (1 H, t, J 9.8, 4b-H), 4.34 (1 H, d, J 3.4, 13b-H), 5.91 (1 H, ddd, J 4.4, 7.8 and 10.3, 6-H), 6.06 (1 H, d, J 2.4, 1-H), 6.34 (1 H, dd, J 2.4 and 8.3, 3-H), 6.55 (1 H, ddd, J 2.9, 9.8 and 10.3, 5-H) and 6.92 (1 H, d, J 8.3, 4-H); $\delta_{\rm C}(68$ MHz) 26.3 (C-9), 28.1 (10-Me), 30.9 (12-Me), 36.0 and 38.1 (C-7 and -13c), 41.7 and 43.5 (C-4b and -8a), 48.2 (C-13b), 55.1 (OMe), 99.4 (C-1), 107.3 and 107.2 (C-3 and -13a), 116.0 (C-4a), 127.9 and 130.0 (C-5 and -6), 136.0 (C-4), 146.9 (C-14a), 149.9 (C-9a), 151.8 (C-11), 158.5 (C-2), 162.4 (C-13) and 208.9 (C-8); m/z 393 (M⁺).

A similar conversion of *endo*-adduct **5e** gave $a \sim 1:1$ mixture of 2,10,12-trimethyl-(**7e-1**) and 4,10,12-trimethyl-*r*-4b,7,8,*c*-8a,9,10,11,12,13,*c*-13b,*c*-13a,14-dodecahydrocyclohepta[*mn*]-



Fig. 2 ORTEP drawing of compound 5b with crystallographic numbering scheme

pyrimido[4,5-c]acridine-8,11,13-trione (7e-2). These isomers could not be separated from each other by the usual procedures. The acridine 7e was obtained as pale yellow prisms from ethanol-chloroform; mp 273-275 °C (decomp.) (Found: C, 70.1; H, 6.1; N, 11.0. C₂₂H₂₃N₃O₃ requires C, 70.01; H, 6.14; N, 11.13%); v_{max}/cm⁻¹ 3350 (NH) and 1690, 1650 and 1640 (CO); m/z 377 (M⁺); $\delta_{\rm H}$ (270 MHz) 2.21 and 2.29 (each 3 H, each s, Me), 2.30–2.42 (total 2 H, ov, 2 × 9-H), 2.60–2.68 (total 2 H, ov, 2×7 -H), 2.74 (1 H, ddd, J 2.2, 4.9 and 19.5, 13c-H) and 2.83 (1 H, ddd, J 3.4, 5.4 and 19.1, 13c-H), 2.93 (1 H, ddd, J 2.9, 4.4 and 19.5, 7-H) and 3.17 (1 H, ddd, J 2.9, 4.9 and 19.1, 7-H), 3.24 (1 H, dd, J 1.0 and 18.1, 9-H) and 3.30 (1 H, dd, J 1.0 and 18.1, 9-H), 3.37, 3.38, 3.45 and 3.51 (each 3 H, each s, 10- and 12-Me), 3.73 (1 H, dd, J 6.0 and 6.8, 8a-H) and 3.78 (1 H, dd, J 5.4 and 6.0, 8a-H), 3.88 (total 2 H, ov, 2 × NH), 4.16 (1 H, d, J 2.9, 13b-H) and 4.30 (1 H, d, J 2.9, 13b-H), 4.23-4.29 (total 2 H, ov, $2 \times 4b$ -H), 5.81–5.93 (2 H, ov, 2×6 -H), 6.37 (1 H, s, 1-H) and 6.47 (1 H, d, J 7.8, 1-H), 6.50-6.59 (total 3 H, ov, ArH), 6.67 (1 H, d, J 7.3, 3-H) and 6.91-6.95 (total 2 H, ov, ArH); $\delta_{\rm C}(68 \text{ MHz})$ 20.6 and 21.1 (Me), 26.5 and 27.0 (C-9), 28.1 and 28.2 (10-Me), 31.0 and 31.7 (12-Me), 35.7, 36.4, 38.4 and 39.6 (C-7 and -13c), 41.0, 41.7, 43.7 and 43.9 (C-4b and -8a), 48.6 and 49.4 (C-13b), 107.0 and 107.4 (C-13a), 113.9 and 115.9 (C-1), 121.0, 121.1, 122.9 and 123.4 (C-3 and -4a), 126.6, 127.5, 128.1 and 129.2 (C-5 and -6), 134.5, 136.0, 136.6 and 136.8 (C-2 and -4), 146.1 and 147.5 (C-14a), 149.9 and 150.1 (C-9a), 152.0 (C-11), 162.5 (C-13) and 208.9 (C-8).

Conversion of exo-adducts 6 into cycloheptaquinazoline 8

General procedures. A solution of *exo*-adduct **6a** (0.100 g, 0.28 mmol) in ethanol (5 cm³) containing three drops of 12 mol dm⁻³ hydrochloric acid was heated under reflux for 24 h. The mixture was evaporated to dryness and the residue was extracted with dichloromethane (3 × 15 cm³). Usual work-up with column chromatography gave the cycloheptaquinazoline **8** (0.021 g, 29%).

Table 4 Crystal data for compounds 5b, 6b and 7f

	5b	6b	7f
Molecular formula	C ₂₁ H ₂₀ BrN ₃ O ₃	C ₂₁ H ₂₀ BrN ₃ O ₃	C ₂₅ H ₂₃ N ₃ O ₃
Relative molecular mass	442.31	442.31	413.48
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	$P2_{1}/c$ (#14)	$P2_{1}/c$ (#14)	$P2_1/c$ (#14)
Cell constants			
<i>a</i> (Å)	11.169(2)	10.790(3)	8.02(2)
$b(\mathbf{A})$	13.513(2)	11.126(4)	11.015(6)
c(Å)	13.471(2)	17.003(2)	22.443(7)
β(°)	108.72(1)	107.16(1)	97.69(6)
$V(Å^3)$	1925.6(4)	1950.2(8)	1964(4)
Z	4	4	4
$D_{\rm c}~({\rm g~cm^{-3}})$	1.526	1.506	1.398



Fig. 3 ORTEP drawing of compound 6b with crystallographic numbering scheme

1,3-Dimethyl-2,3,4,10-tetrahydro-1H-cyclohepta[g]quinazoline-2,4,10-trione **8** was obtained as yellow needles from ethanol-chlorofom; mp 210–212 °C (Found: C, 67.0; H, 4.6; N, 10.2. $C_{15}H_{12}N_2O_3$ requires C, 67.15; H, 4.51; N, 10.44%); v_{max} /cm⁻¹ 1700 and 1650 (CO); δ_H (270 MHz) 3.53 and 3.74 (each 3 H, each s, 1- and 3-Me), 6.74 (1 H, ddd, J 1.1, 8.1 and 11.4, 7-H), 6.97 (1 H, dd, J 1.1 and 12.1, 9-H), 7.16 (1 H, ddd, J 1.1, 8.1 and 12.1, 8-H), 7.44 (1 H, d, J 11.4, 6-H), 8.35 (1 H, s, 5-H) and 8.58 (1 H, s, 11-H); δ_C (68 MHz) 28.8 (1-Me), 31.3 (3-Me), 115.5 (C-8), 118.4 (C-4a), 126.1 (C-6), 130.8 (C-5a), 134.9, 136.1 and 136.8 (C-5, -7, and -11), 138.7 (C-9), 140.7 (C-10a), 143.0 (C-11a), 150.9 (C-2), 161.1 (C-4) and 187.1 (C-10); m/z 268 (M⁺).

Single-crystal X-ray structure determinations

Single crystals (prisms) of compound **5b**, **6b** and **7f** for X-ray diffraction studies were recrystallised from ethanol. A crystal of approximate dimensions of $0.240 \times 0.460 \times 0.540$ mm was used for data collection of compound **5b**, one of $0.240 \times 0.240 \times 0.260$ mm of compound **6b** and one of $0.040 \times 0.0400 \times 0.460$ mm of compound **7f**. All measure-



Fig. 4 ORTEP drawing of compound 7f with crystallographic numbering scheme

ments were made on a Rigaku AFC5S diffractometer by employing graphite-monochromated Mo-Ka radiation. The unit-cell dimensions were obtained by least-squares analysis of 24 or 25 reflections within the range $37.25 < 2\theta < 39.69^{\circ}$ for compound 5b, $34.49 < 2\theta < 39.88^{\circ}$ for compound 6b and $20.24 < 2\theta < 25.22$ for compound **7f**, respectively. Summaries of the crystal data for compound 5b, 6b and 7f are given in Table 4. The ω -2 θ scan technique to a maximum 2 θ -value of 55° was used. Scans of $(1.42 + 0.30 \tan \theta)^{\circ}$ were made at a speed of $32.0^{\circ} \text{ min}^{-1}$ (in omega) for compound **5b**, of (0.79 + 0.30 tan θ)° at a speed of 32.0° min⁻¹ for compound **6b**, and of (1.00 + 0.30 tan θ)° at a speed of 32.0° min⁻¹ for compound 7f. A total of 4848 observed reflections (unique: 4620; R_{int} 0.082) for compound **5b**, 4955 (unique: 4711; *R*_{int} 0.063) for compound **6b**, and 5098 (unique: 4766; R_{int} 0.127) for compound 7f were collected. All calculations were performed using the TEXSAN program.8 Atoms other than hydrogen were refined anisotropically. The structures were solved by direct methods (MITHRIL)⁹ and refined by least squares to R 0.072 (compound 5b), 0.059 (compound 6b) and 0.061 (compound 7f). ORTEP¹⁰ drawings of compounds 5b, 6b and 7f are shown in Figs. 2, 3 and 4, respectively.[‡]

[‡] Tables of atomic coordinates, bond lengths and bond angles have been deposited at the Cambridge Crystallographic Data Centre. See *Instructions for Authors* (1995), Issue 1. **Computational procedure.** The structural data of compounds **5b** and **7f**, confirmed by X-ray structure analyses, were used as initial geometries, and the structures of the parent heterocycles **5a** and **7a** were fully optimerised individually by the PM3 method¹¹ using the MOPAC program (Version 6.00)¹² on the VAX 4000 in Ube Laboratory, Corporate Research & Development, Ube Industries Ltd. The calculated heats of formation are summarised in Fig. 1.

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